Photochromism of triangle terthiophene derivatives as molecular re-router†

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2,2'-3,3"-Terthiophene derivatives undergo photochemically **reversible cyclization and cycloreversion reactions. The absorption peak wavelength changed systematically with substitution** of the phenyl rings at 5-, 5'- and 5"-positions of the thiophene rings, which indicates re-routing of the π -conjugation system.

Photochromic molecules have been widely studied as photoswitching units in various photoresponsive molecules and polymers.1 Recently, photoswitching effects in photochromic diarylethenes have been extensively studied for controlling various chemical and physical properties such as refractive index, dielectric properties, electronic conduction, electrochemical response and magnetic interactions.2–9 Some of these photoswitching effects are based, at least partly, on changes in the extent of the π -conjugation in diarylethene upon the photochromic reactions. That is, the π conjugation system of diarylethene extends over both sides of the molecule in the closed-ring form, while it is restricted on each side of its open-ring isomer, as shown in Scheme 1. The connection mode of the π -conjugation system is dependent on the substitution positions of the thiophene units to the ethene unit.10 Recently, Branda *et al.* have designed dithienylethene derivatives having thiophene units on their reacting carbon atoms for a photoswitching unit of the π -conjugation pathway, although various types of photocyclization reactions are possible in these molecules and the detailed reaction mechanism must be clarified.11 Since various types of molecular electronic devices have been designed on the basis of a π -conjugation connection pathway,¹² photon-mode switching and re-routing of the connection mode of the π conjugation system are worthy of extensive study.

Here, we report on photochromic triangle terthiophene derivatives **1–3**, which behave as photoswitching 2-way re-routers of the π -conjugation system. These molecules include a hexatriene unit in their molecular structure and are expected to undergo photoinduced cyclization and cycloreversion reactions in a similar manner to diarylethenes. As illustrated in Scheme 2, the π -conjugation system in the open-ring isomers would extend between the 5- and 5'positions in the central and left side thiophene rings, respectively, while it would extend between the $5-$ and $5''$ -positions in the central and right side thiophene rings in the closed-ring form isomer, respectively. In order to demonstrate the re-routing effect in the π conjugation connection mode, molecules **1–3** having different substituents were prepared.

Compounds **1a–3a** were synthesized by conventional crosscoupling reactions of thiophene derivatives and their chemical structures were confirmed by elemental analysis, mass spectra and NMR spectra.† The closed-ring isomers **1b–3b** were prepared by

† Electronic supplementary information (ESI) available: analytical data for **1a**, **2a** and **3a**. See http://www.rsc.org/suppdata/cc/b3/b311334e/

irradiating hexane solutions of the corresponding open ring isomers **1a–3a** with UV light and were isolated by HPLC from the colored solutions. Formation of the colored isomers was confirmed by mass spectrometry and/or by $H-MMR$ spectroscopy in CDCl₃ solution. The photochromic reaction behavior and absorption spectral changes of these molecules were studied in hexane at room temperature.

Colorless and yellow solutions of the open ring isomers **1a–3a** were observed to turn red or blue depending on the molecular structures upon irradiation with UV light (λ = 313 nm). These colors disappeared upon irradiation with visible light ($\lambda > 460$ nm). These coloration and bleaching cycles can be repeated many times with alternating irradiation with UV (λ = 313 nm) and visible light ($\lambda > 460$ nm).

As shown in Fig. 1, **1a** has no absorption band in the visible range and a new absorption band appeared at 558 nm upon irradiation with UV light, which corresponds to the formation of **1b**. An isosbestic point was observed at 318 nm supporting the twocomponent photochromic reactions shown in Scheme 2. The conversion ratio from **1a** to **1b** was 72% at the photostationary state which was achieved by irradiation with UV light ($\lambda = 313$ nm). The colored solution was completely bleached upon irradiation with visible light. Similar photochromic performance was also observed for **2** and **3**. Fig. 1(b) and (c) show absorption spectra of **1a–3a** and **1b–3b**, respectively. Their λ_{max} and ϵ_{max} are summarized in Table 1. The λ_{max} of the open-ring isomers showed a systematic red shift from **1a** and **3a**. This is due to the introduction of phenyl groups at the 5- and 5'-positions of the thiophene rings. These spectral shifts of indicate that the π -conjugation pathway is connected between the 5- and 5'-positions and is disconnected

between the 5- and 5"-positions as schematically illustrated in Scheme 2.

On the other hand, λ_{max} of the closed ring isomers showed a red shift in the absorption maximum upon the introduction of a phenyl ring at the 5-position of the central thiophene ring. Almost identical absorption spectra were observed for **2b** and **3b**. Substitution with a phenyl unit on the 5'-position does not affect the absorption spectrum. These spectral features indicate that the π -conjugation systems are connected between the 5- and the 5"-positions of the closed ring isomers but those between the 5- and 5'-positions are disconnected.

Similar spectral shifts were also observed in fluorescence emission wavelengths, as also listed in Table 1. Noted that the closed-ring isomers **1b** and **2b** showed fluorescence emission wavelengths different from those of **1a** and **2a**, respectively. That is, the emission wavelength can be modulated by the photochromic reaction, although their emission intensities were relatively weak as found for some diarylethene derivatives.13

Fig. 1 (a) Absorption spectra of **1a** solution (solid line) and of the photostationary state under irradiation with UV light ($\lambda = 313$ nm) (dotted line) and absorption spectrum of **1b** solution (broken line). (b) Absorption spectra of open ring isomers **1a–3a**. (c) Absorption spectra of closed ring isomers **1b–3b**. All measurements were carried out at room temp. using hexane as solvent.

Table 1 Optical properties of photochromic terthiophenes **1–3***a*

Compound	$\lambda_{\rm max}/\rm nm$ $(\varepsilon/M^{-1}$ cm ⁻¹)	$\phi_{\alpha-c}{}^b$	ϕ_{c-o} ^c	$\lambda_{\rm em}$, $\lambda_{\rm ex}/\rm nm$
1a	264 (20900)	0.27		389, 313
	294 (21500)			
1 _b	558 (10100)		0.43	561, 682
2a	272 (31500)	0.18		432, 348
	307 (28100)			
2 _b	584 (16000)		0.26	716, 586
3a	274 (31500)	0.04		449, 352
	308 (27000)			
	345 (26000)			
3 _b	584 (16000)		0.39	
	^{<i>a</i>} All measurements were performed at room temp. in hexane solution. ^{<i>b</i>} At			
	313 nm. ϵ At λ_{max} of the closed ring isomers.			

Table 1 also summarizes the photochemical quantum yields of cyclization and cycloreversion reactions of the compounds. These values were evaluated by the standard procedure using fulgide as a standard.14 **1** showed relatively high photochemical quantum yields in both cyclization and cycloreversion reactions. Suppressed quantum yields were observed in **3a**. No marked thermal bleaching was observed in the closed ring isomers **1b**, **2b** and **3b** at room temp.

In conclusion, the triangular terthiophene derivatives presented here showed reversible photochromic reactions with relatively high photochemical quantum yields and they can re-route the connection mode of the π -conjugation system with the photochromic reaction. We expect various kinds of photo-switching effects such as electrical conductivity and optical properties with these photoswitching units. By combining these molecular re-routers, photonmode logic gates may be realized in the future.

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